

CHEMISTRY & CHEMICAL BIOLOGY COLLOQUIUM: Spectroscopic Probes of Plasmon-Driven Chemical Reactions

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<u>Date:</u> 10/9/2020

<u>Time:</u> 1:30 PM-2:50 PM

<u>Link:</u>

Please email snsgradstaff@ucmerced.edu for Zoom link and passcode



About The Speaker:

Renee R. Frontiera is a McKnight Land-Grant associate professor of Chemistry at the University of Minnesota. Her research group uses Raman spectroscopic techniques to examine chemical composition and chemical reaction dynamics on nanometer length scales and ultrafast time scales. She received her Ph. D. in 2009 from the University of California – Berkeley, under the advisement of Richard A. Mathies. Her postdoctoral research at Northwestern University was under the supervision of Richard P. Van Duyne. Her research group at the University of Minnesota was founded in 2013, and she is the recent recipient of an NSF CAREER award, a DOE Early Career award, and an NIH Maximizing Investigators' Research Award (MIRA). She was named one of Chemical & Engineering News's "Talented 12", and has won a Journal of Physical Chemistry Lectureship and a Camille Dreyfus Teacher-Scholar award.

Abstract:

Plasmonic materials are highly promising photoredox catalysts for driving energetically unfavorable chemical reactions with light, due to their large optical cross sections and ability to generate a number of hot holes and electrons. However, the efficiencies of most plasmondriven processes are quite low, likely due to the lack of mechanistic understanding of the underlying physical processes. Plasmons can concentrate electromagnetic fields, can generate highly energetic electrons and holes, and can heat up local environments. An understanding of the energy partitioning into each of these processes is crucial to the design of plasmonic photocatalysts which are optimized for chemical selectivity. Here I'll discuss our development of ultrafast surface-enhanced Raman spectroscopy (SERS) to probe the behavior of molecules in plasmonic hot spots. By probing on the relevant timescales, we are able to uncover how and when the plasmon energy is converted into hot carriers, molecular heating, or resonant energy transfer. We use a new form of ultrafast Raman thermometry to probe energy transferred into adsorbed molecules as heat, showing that the plasmoninduced heating contribution to catalysis is negligible. Additionally, we probe hot carrier transfer events and use SERS substrates with tunable electromagnetic field enhancements to examine the effects of local fields on chemical reactivity. These results quantitatively determine the energy partitioning of plasmon decay, and highlight the unique ability of these plasmonic materials to drive energetically unfavorable chemical reactions.

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